

# 1-[(3-Methylpiperidin-1-yl)(phenyl)-methyl]-2-naphthol

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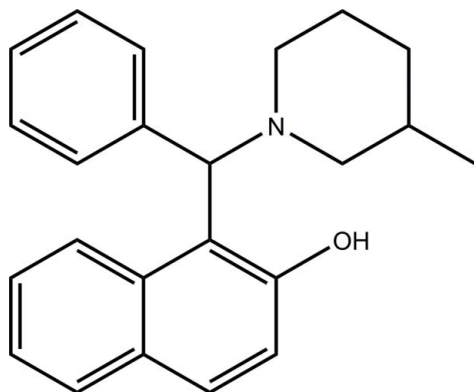
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Key indicators: single-crystal X-ray study;  $T = 292$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.061;  $wR$  factor = 0.155; data-to-parameter ratio = 19.4.

In the title compound,  $\text{C}_{23}\text{H}_{25}\text{NO}$ , the dihedral angle between the naphthylene ring system and the benzene ring is  $78.17(10)^\circ$ . The molecular conformation is stabilized by a strong intramolecular  $\text{O}-\text{H}\cdots\text{N}$  hydrogen bond.

## Related literature

For the structures of related compounds, see: Szatmari & Fulop (2004); Zhao & Sun (2005); Wang & Zhao (2008); Wan & Zhao (2008).



## Experimental

### Crystal data

$\text{C}_{23}\text{H}_{25}\text{NO}$	$V = 1858.5(5) \text{ \AA}^3$
$M_r = 331.44$	$Z = 4$
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation
$a = 9.2138(13) \text{ \AA}$	$\mu = 0.07 \text{ mm}^{-1}$
$b = 10.9056(13) \text{ \AA}$	$T = 292 \text{ K}$
$c = 18.635(3) \text{ \AA}$	$0.30 \times 0.25 \times 0.20 \text{ mm}$
$\beta = 97.007(10)^\circ$	

### Data collection

Rigaku SCXmini diffractometer	19217 measured reflections
Absorption correction: multi-scan ( <i>CrystalClear</i> ; Rigaku, 2005)	4422 independent reflections
$T_{\min} = 0.965$ , $T_{\max} = 0.979$	2302 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.067$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.061$	228 parameters
$wR(F^2) = 0.155$	H-atom parameters constrained
$S = 0.99$	$\Delta\rho_{\max} = 0.15 \text{ e \AA}^{-3}$
4422 reflections	$\Delta\rho_{\min} = -0.16 \text{ e \AA}^{-3}$

**Table 1**

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{O1}-\text{H1A}\cdots\text{N1}$	0.82	1.84	2.570 (2)	148

Data collection: *CrystalClear* (Rigaku, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL/PC* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL/PC*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: RZ2319).

## References

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**supplementary materials**

*Acta Cryst.* (2009). E65, o1277 [ doi:10.1107/S1600536809017176 ]

## 1-[(3-Methylpiperidin-1-yl)(phenyl)methyl]-2-naphthol

W. X. Wang and H. Zhao

### Comment

It is well known that many compounds derived from naphthalen-2-ol have received much attention in organic chemistry (Szatmari & Fulop, 2004; Zhao & Sun, 2005). Recently, we reported the synthesis and crystal structures of 1-[(dimethylamino)(phenyl)methyl]naphthalen-2-ol (Wang & Zhao, 2008) and 1-[pyrrolidin-1-yl(*p*-tolyl)methyl]naphthalen-2-ol (Wan & Zhao, 2008). We now report the crystal structure of the title compound.

Bond lengths and angles in the title compound have normal values. The dihedral angle between the naphthyl and phenyl rings is 78.17 (10)°. The molecular conformation is stabilized by a strong intramolecular O—H···N hydrogen bond (Table 1). The crystal packing is mainly stabilized by van der Waals interactions.

### Experimental

A dry 50 ml flask was charged with benzaldehyde (10 mmol), naphthalen-2-ol (10 mmol) and 3-methylpiperidine (10 mmol). The mixture was stirred at 373 K for 10 h, then ethanol (15 ml) was added. After heating under reflux for 30 min, the precipitate was filtrated off and washed 3 times with ethanol to give the title compound. Single crystals suitable for X-ray analysis were obtained by slow evaporation of a dichloromethane solution.

### Refinement

All H atoms were calculated geometrically, with C—H = 0.93–0.98 Å, O—H = 0.82 Å, and refined as riding with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  or  $1.2U_{\text{eq}}(\text{C}, \text{O})$  for methyl and hydroxy Y atoms.

### Figures

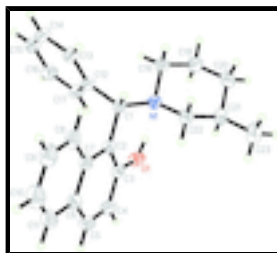


Fig. 1. The structure of the title compound, showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level.

## 1-[(3-methylpiperidin-1-yl)(phenyl)methyl]-2-naphthol

*Crystal data*

C<sub>23</sub>H<sub>25</sub>NO

$F_{000} = 712$

# supplementary materials

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$M_r = 331.44$

Monoclinic,  $P2_1/n$

Hall symbol: -P 2yn

$a = 9.2138$  (13) Å

$b = 10.9056$  (13) Å

$c = 18.635$  (3) Å

$\beta = 97.007$  (10)°

$V = 1858.5$  (5) Å<sup>3</sup>

$Z = 4$

$D_x = 1.185$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation

$\lambda = 0.71073$  Å

Cell parameters from 2989 reflections

$\theta = 2.6$ – $27.9$ °

$\mu = 0.07$  mm<sup>-1</sup>

$T = 292$  K

Prism, pale yellow

$0.30 \times 0.25 \times 0.20$  mm

## Data collection

Rigaku SCXmini  
diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

Detector resolution: 13.6612 pixels mm<sup>-1</sup>

$T = 292$  K

$\omega$  scans

Absorption correction: multi-scan  
(CrystalClear; Rigaku, 2005)

$T_{\min} = 0.965$ ,  $T_{\max} = 0.979$

19217 measured reflections

4422 independent reflections

2302 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.067$

$\theta_{\max} = 27.9$ °

$\theta_{\min} = 2.9$ °

$h = -12 \rightarrow 12$

$k = -14 \rightarrow 14$

$l = -24 \rightarrow 24$

## Refinement

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.061$

$wR(F^2) = 0.155$

$S = 0.99$

4422 reflections

228 parameters

Primary atom site location: structure-invariant direct  
methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring  
sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0693P)^2]$$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.15$  e Å<sup>-3</sup>

$\Delta\rho_{\min} = -0.16$  e Å<sup>-3</sup>

Extinction correction: none

## Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -

factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.67985 (19)	0.98342 (18)	0.20764 (10)	0.0429 (5)
H1	0.6443	1.0644	0.2203	0.051*
C2	0.78426 (19)	0.93769 (17)	0.27180 (10)	0.0424 (5)
C3	0.9016 (2)	0.86366 (18)	0.26209 (11)	0.0473 (5)
C4	0.9972 (2)	0.8201 (2)	0.32150 (13)	0.0588 (6)
H4	1.0778	0.7728	0.3138	0.071*
C5	0.9718 (2)	0.8470 (2)	0.38980 (13)	0.0640 (6)
H5	1.0349	0.8168	0.4285	0.077*
C6	0.8527 (2)	0.9195 (2)	0.40364 (11)	0.0560 (6)
C7	0.7590 (2)	0.96718 (17)	0.34408 (10)	0.0466 (5)
C8	0.6424 (2)	1.04328 (19)	0.36039 (12)	0.0580 (6)
H8	0.5790	1.0764	0.3227	0.070*
C9	0.6211 (3)	1.0690 (2)	0.42986 (13)	0.0739 (7)
H9	0.5445	1.1201	0.4388	0.089*
C10	0.7124 (3)	1.0197 (3)	0.48762 (14)	0.0845 (9)
H10	0.6959	1.0367	0.5349	0.101*
C11	0.8255 (3)	0.9467 (2)	0.47479 (12)	0.0724 (7)
H11	0.8863	0.9139	0.5136	0.087*
C12	0.5482 (2)	0.89924 (19)	0.19333 (10)	0.0466 (5)
C13	0.4093 (2)	0.9444 (2)	0.19799 (11)	0.0601 (6)
H13	0.3973	1.0267	0.2090	0.072*
C14	0.2885 (2)	0.8692 (3)	0.18655 (13)	0.0719 (7)
H14	0.1959	0.9008	0.1901	0.086*
C15	0.3047 (3)	0.7479 (3)	0.16990 (13)	0.0713 (7)
H15	0.2231	0.6972	0.1620	0.086*
C16	0.4413 (2)	0.7016 (2)	0.16489 (11)	0.0655 (6)
H16	0.4524	0.6193	0.1536	0.079*
C17	0.5626 (2)	0.7767 (2)	0.17657 (11)	0.0541 (5)
H17	0.6550	0.7444	0.1731	0.065*
C18	0.6619 (2)	1.0167 (2)	0.07506 (10)	0.0580 (6)
H18A	0.6042	1.0904	0.0784	0.070*
H18B	0.5954	0.9477	0.0671	0.070*
C19	0.7514 (2)	1.0277 (2)	0.01231 (11)	0.0680 (7)
H19A	0.8024	0.9511	0.0068	0.082*
H19B	0.6864	1.0421	-0.0318	0.082*
C20	0.8616 (2)	1.1307 (2)	0.02313 (11)	0.0634 (6)
H20A	0.8106	1.2087	0.0216	0.076*
H20B	0.9225	1.1298	-0.0158	0.076*
C21	0.9571 (2)	1.11783 (19)	0.09503 (11)	0.0531 (5)
H21	1.0152	1.0428	0.0935	0.064*
C22	0.8604 (2)	1.10436 (19)	0.15507 (11)	0.0509 (5)
H22A	0.9217	1.0940	0.2008	0.061*

## supplementary materials

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H22B	0.8038	1.1788	0.1580	0.061*
C23	1.0612 (3)	1.2248 (2)	0.11110 (14)	0.0819 (8)
H23A	1.0064	1.2996	0.1113	0.123*
H23B	1.1262	1.2293	0.0746	0.123*
H23C	1.1171	1.2132	0.1576	0.123*
N1	0.75966 (16)	0.99877 (14)	0.14314 (8)	0.0455 (4)
O1	0.93357 (14)	0.82835 (13)	0.19587 (8)	0.0575 (4)
H1A	0.8851	0.8690	0.1646	0.086*

### Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.0391 (10)	0.0487 (11)	0.0419 (11)	0.0029 (9)	0.0094 (8)	-0.0008 (9)
C2	0.0363 (10)	0.0451 (10)	0.0454 (11)	-0.0067 (9)	0.0030 (8)	-0.0013 (9)
C3	0.0389 (11)	0.0505 (12)	0.0522 (13)	-0.0056 (9)	0.0047 (9)	-0.0016 (10)
C4	0.0405 (11)	0.0570 (13)	0.0758 (17)	-0.0008 (10)	-0.0054 (11)	0.0039 (12)
C5	0.0608 (15)	0.0662 (15)	0.0594 (16)	-0.0155 (13)	-0.0149 (11)	0.0122 (12)
C6	0.0570 (13)	0.0584 (14)	0.0507 (13)	-0.0203 (12)	-0.0008 (10)	0.0018 (11)
C7	0.0485 (12)	0.0482 (11)	0.0435 (12)	-0.0139 (10)	0.0070 (9)	-0.0010 (9)
C8	0.0602 (14)	0.0652 (14)	0.0513 (13)	-0.0089 (12)	0.0177 (10)	-0.0074 (11)
C9	0.0778 (17)	0.0849 (18)	0.0635 (17)	-0.0153 (15)	0.0265 (14)	-0.0185 (14)
C10	0.100 (2)	0.105 (2)	0.0508 (16)	-0.0353 (18)	0.0194 (15)	-0.0203 (15)
C11	0.0845 (18)	0.0852 (18)	0.0454 (14)	-0.0310 (16)	-0.0009 (12)	0.0018 (13)
C12	0.0365 (10)	0.0610 (13)	0.0423 (11)	0.0007 (10)	0.0045 (8)	0.0067 (10)
C13	0.0417 (12)	0.0763 (15)	0.0633 (14)	0.0043 (12)	0.0100 (10)	0.0097 (12)
C14	0.0362 (12)	0.109 (2)	0.0708 (16)	-0.0001 (14)	0.0087 (11)	0.0197 (15)
C15	0.0502 (14)	0.095 (2)	0.0660 (15)	-0.0220 (14)	-0.0047 (11)	0.0137 (14)
C16	0.0624 (15)	0.0728 (16)	0.0578 (14)	-0.0103 (13)	-0.0066 (11)	-0.0003 (12)
C17	0.0418 (11)	0.0633 (13)	0.0557 (13)	-0.0011 (11)	0.0001 (9)	-0.0021 (11)
C18	0.0482 (12)	0.0814 (16)	0.0437 (12)	0.0037 (11)	0.0036 (10)	0.0040 (11)
C19	0.0598 (14)	0.1003 (19)	0.0444 (13)	-0.0009 (14)	0.0082 (10)	0.0022 (12)
C20	0.0637 (14)	0.0794 (16)	0.0503 (13)	0.0048 (13)	0.0201 (11)	0.0082 (12)
C21	0.0520 (12)	0.0551 (12)	0.0551 (13)	-0.0024 (10)	0.0184 (10)	-0.0001 (10)
C22	0.0498 (12)	0.0539 (12)	0.0500 (12)	-0.0030 (10)	0.0105 (9)	-0.0017 (10)
C23	0.0856 (18)	0.0819 (17)	0.0834 (18)	-0.0242 (15)	0.0308 (14)	-0.0037 (14)
N1	0.0394 (9)	0.0572 (10)	0.0403 (9)	-0.0036 (8)	0.0065 (7)	-0.0003 (8)
O1	0.0443 (8)	0.0662 (10)	0.0620 (10)	0.0069 (7)	0.0062 (7)	-0.0071 (8)

### Geometric parameters ( $\text{\AA}$ , $^\circ$ )

C1—N1	1.493 (2)	C14—H14	0.9300
C1—C12	1.518 (3)	C15—C16	1.370 (3)
C1—C2	1.524 (3)	C15—H15	0.9300
C1—H1	0.9800	C16—C17	1.381 (3)
C2—C3	1.379 (3)	C16—H16	0.9300
C2—C7	1.431 (3)	C17—H17	0.9300
C3—O1	1.359 (2)	C18—N1	1.477 (2)
C3—C4	1.411 (3)	C18—C19	1.516 (3)
C4—C5	1.354 (3)	C18—H18A	0.9700

C4—H4	0.9300	C18—H18B	0.9700
C5—C6	1.402 (3)	C19—C20	1.510 (3)
C5—H5	0.9300	C19—H19A	0.9700
C6—C11	1.410 (3)	C19—H19B	0.9700
C6—C7	1.419 (3)	C20—C21	1.517 (3)
C7—C8	1.419 (3)	C20—H20A	0.9700
C8—C9	1.362 (3)	C20—H20B	0.9700
C8—H8	0.9300	C21—C23	1.517 (3)
C9—C10	1.391 (4)	C21—C22	1.520 (3)
C9—H9	0.9300	C21—H21	0.9800
C10—C11	1.356 (4)	C22—N1	1.479 (2)
C10—H10	0.9300	C22—H22A	0.9700
C11—H11	0.9300	C22—H22B	0.9700
C12—C17	1.383 (3)	C23—H23A	0.9600
C12—C13	1.384 (3)	C23—H23B	0.9600
C13—C14	1.378 (3)	C23—H23C	0.9600
C13—H13	0.9300	O1—H1A	0.8200
C14—C15	1.371 (3)		
N1—C1—C12	112.81 (15)	C14—C15—H15	120.1
N1—C1—C2	110.03 (15)	C15—C16—C17	120.2 (2)
C12—C1—C2	110.74 (15)	C15—C16—H16	119.9
N1—C1—H1	107.7	C17—C16—H16	119.9
C12—C1—H1	107.7	C16—C17—C12	120.7 (2)
C2—C1—H1	107.7	C16—C17—H17	119.6
C3—C2—C7	118.38 (18)	C12—C17—H17	119.6
C3—C2—C1	121.25 (17)	N1—C18—C19	109.91 (16)
C7—C2—C1	120.33 (17)	N1—C18—H18A	109.7
O1—C3—C2	123.06 (18)	C19—C18—H18A	109.7
O1—C3—C4	115.67 (18)	N1—C18—H18B	109.7
C2—C3—C4	121.3 (2)	C19—C18—H18B	109.7
C5—C4—C3	120.1 (2)	H18A—C18—H18B	108.2
C5—C4—H4	120.0	C20—C19—C18	112.10 (19)
C3—C4—H4	120.0	C20—C19—H19A	109.2
C4—C5—C6	121.6 (2)	C18—C19—H19A	109.2
C4—C5—H5	119.2	C20—C19—H19B	109.2
C6—C5—H5	119.2	C18—C19—H19B	109.2
C5—C6—C11	121.6 (2)	H19A—C19—H19B	107.9
C5—C6—C7	118.5 (2)	C19—C20—C21	110.91 (17)
C11—C6—C7	119.9 (2)	C19—C20—H20A	109.5
C6—C7—C8	116.79 (19)	C21—C20—H20A	109.5
C6—C7—C2	120.03 (19)	C19—C20—H20B	109.5
C8—C7—C2	123.18 (19)	C21—C20—H20B	109.5
C9—C8—C7	121.6 (2)	H20A—C20—H20B	108.0
C9—C8—H8	119.2	C20—C21—C23	112.82 (18)
C7—C8—H8	119.2	C20—C21—C22	109.29 (17)
C8—C9—C10	120.8 (3)	C23—C21—C22	110.02 (17)
C8—C9—H9	119.6	C20—C21—H21	108.2
C10—C9—H9	119.6	C23—C21—H21	108.2
C11—C10—C9	119.7 (2)	C22—C21—H21	108.2

## supplementary materials

C11—C10—H10	120.1	N1—C22—C21	112.21 (16)
C9—C10—H10	120.1	N1—C22—H22A	109.2
C10—C11—C6	121.2 (2)	C21—C22—H22A	109.2
C10—C11—H11	119.4	N1—C22—H22B	109.2
C6—C11—H11	119.4	C21—C22—H22B	109.2
C17—C12—C13	118.2 (2)	H22A—C22—H22B	107.9
C17—C12—C1	121.84 (17)	C21—C23—H23A	109.5
C13—C12—C1	119.93 (19)	C21—C23—H23B	109.5
C14—C13—C12	121.0 (2)	H23A—C23—H23B	109.5
C14—C13—H13	119.5	C21—C23—H23C	109.5
C12—C13—H13	119.5	H23A—C23—H23C	109.5
C15—C14—C13	120.1 (2)	H23B—C23—H23C	109.5
C15—C14—H14	120.0	C18—N1—C22	109.40 (15)
C13—C14—H14	120.0	C18—N1—C1	113.43 (14)
C16—C15—C14	119.8 (2)	C22—N1—C1	109.15 (14)
C16—C15—H15	120.1	C3—O1—H1A	109.5
N1—C1—C2—C3	-31.0 (2)	C7—C6—C11—C10	-1.1 (3)
C12—C1—C2—C3	94.4 (2)	N1—C1—C12—C17	63.6 (2)
N1—C1—C2—C7	151.44 (16)	C2—C1—C12—C17	-60.2 (2)
C12—C1—C2—C7	-83.2 (2)	N1—C1—C12—C13	-117.4 (2)
C7—C2—C3—O1	178.73 (17)	C2—C1—C12—C13	118.81 (19)
C1—C2—C3—O1	1.1 (3)	C17—C12—C13—C14	0.3 (3)
C7—C2—C3—C4	-1.6 (3)	C1—C12—C13—C14	-178.80 (18)
C1—C2—C3—C4	-179.21 (17)	C12—C13—C14—C15	-0.3 (3)
O1—C3—C4—C5	-177.90 (18)	C13—C14—C15—C16	0.2 (4)
C2—C3—C4—C5	2.4 (3)	C14—C15—C16—C17	0.0 (3)
C3—C4—C5—C6	-0.9 (3)	C15—C16—C17—C12	-0.1 (3)
C4—C5—C6—C11	179.4 (2)	C13—C12—C17—C16	0.0 (3)
C4—C5—C6—C7	-1.3 (3)	C1—C12—C17—C16	179.00 (18)
C5—C6—C7—C8	-178.04 (18)	N1—C18—C19—C20	-57.0 (3)
C11—C6—C7—C8	1.2 (3)	C18—C19—C20—C21	53.8 (3)
C5—C6—C7—C2	2.0 (3)	C19—C20—C21—C23	-175.32 (19)
C11—C6—C7—C2	-178.67 (17)	C19—C20—C21—C22	-52.6 (2)
C3—C2—C7—C6	-0.6 (3)	C20—C21—C22—N1	57.4 (2)
C1—C2—C7—C6	177.01 (17)	C23—C21—C22—N1	-178.25 (18)
C3—C2—C7—C8	179.47 (17)	C19—C18—N1—C22	59.5 (2)
C1—C2—C7—C8	-2.9 (3)	C19—C18—N1—C1	-178.37 (18)
C6—C7—C8—C9	-0.2 (3)	C21—C22—N1—C18	-61.3 (2)
C2—C7—C8—C9	179.68 (19)	C21—C22—N1—C1	174.09 (16)
C7—C8—C9—C10	-0.9 (3)	C12—C1—N1—C18	43.3 (2)
C8—C9—C10—C11	1.1 (4)	C2—C1—N1—C18	167.50 (16)
C9—C10—C11—C6	0.0 (4)	C12—C1—N1—C22	165.56 (15)
C5—C6—C11—C10	178.1 (2)	C2—C1—N1—C22	-70.23 (19)

### Hydrogen-bond geometry ( $\text{\AA}$ , $^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O1—H1A $\cdots$ N1	0.82	1.84	2.570 (2)	148



Fig. 1

